

**COURSE DATA****Data Subject**

<b>Code</b>	45011
<b>Name</b>	Química Sostenible
<b>Cycle</b>	Master's degree
<b>ECTS Credits</b>	3.0
<b>Academic year</b>	2022 - 2023

**Study (s)**

<b>Degree</b>	<b>Center</b>	<b>Acad. Period year</b>
2249 - Master's Degree in Chemistry	Faculty of Chemistry	1 First term

**Subject-matter**

<b>Degree</b>	<b>Subject-matter</b>	<b>Character</b>
2249 - Master's Degree in Chemistry	7 - Optatividad en Química	Optional

**Coordination**

<b>Name</b>	<b>Department</b>
MUÑOZ ESPI, RAFAEL	315 - Physical Chemistry

**SUMMARY**

Green Chemistry is the orientation of chemistry, as a set of theoretical and applied knowledge, whose specific aim is the prevention of environmental contamination and the risks associated with chemical products, by introducing or stressing of clean and safe production processes and of less toxic and contaminant chemical products without reducing its contribution to wellness and technological progress.

Sustainable Chemistry must be considered as a part of Environmental Chemistry and endeavours to achieve the current and future prevention of contamination and risk problems originated by chemical substances, by analysing the origin of these problems. With the premises, the objectives of Green Chemistry are the following:

Reducing the generation and use of contaminants in the chemical process

Reducing the risky character of the chemical process



Reducing the noxious effect of the chemicals used by the production sectors or the final consumer.

Reducing the use of extinguishable and scarce raw materials.

## PREVIOUS KNOWLEDGE

### Relationship to other subjects of the same degree

There are no specified enrollment restrictions with other subjects of the curriculum.

### Other requirements

Chemistry knowledge acquired during the Chemistry Degree are required.

## COMPETENCES (RD 1393/2007) // LEARNING OUTCOMES (RD 822/2021)

### 2249 - Master's Degree in Chemistry

- Students should apply acquired knowledge to solve problems in unfamiliar contexts within their field of study, including multidisciplinary scenarios.
- Be able to solve complex chemistry problems, whether in the academic, research or industrial application areas at a specialization or masters-level.
- Possess the necessary skills to develop multidisciplinary activities within the field of chemistry at the master's level.
- Be able to design, conduct, analyse and interpret complex experiments and data, as a specialist.
- Apply the advanced theoretical and practical knowledge gained in the different specialties of chemistry to R&D and innovation.

## LEARNING OUTCOMES (RD 1393/2007) // NO CONTENT (RD 822/2021)

- Be able to approach from the experimental and theoretical point of view real problems of a specialized scientific and / or technological nature, as well as propose solutions, in different areas of Chemistry.
- Be able to function in professional scientific-technological environments related to industry, research, development and / or innovation.
- Be able to transmit and disseminate results of scientific-technological activity.
- Be able to valorise the role of Chemistry and its influence on the environment
- Be able to valorise the important role of Green Chemistry in the search of more efficient and environmental friendly products and processes.
- Identify the residues generated in the different steps of the chemical processes in order to proceed to its possible reuse or selective collection of toxic residues.
- Describe the main sources of chemical products and their treatment for the transformation into added value materials.
- Know the role of chemistry in the main sources of renewable energy and in the methods for energy storage.



- Know the recycling processes of the main materials and metals.
- Design, synthesise and perform effective analytical processes for obtaining and valorising products.
- Use the different sustainable tools of Chemistry.
- Knowing how to apply the knowledge acquired to contribute to the Sustainable Development Goals (SDGs), such as the sustainable management of water, raw materials and energy sources (SDGs 6 and 7) and to develop a professional work with the least environmental impact and using alternative raw materials (SDGs 11, 14 and 15)

## DESCRIPTION OF CONTENTS

### 1. Introduction

Objectives. Use of renewable sources of raw materials. Reduction of polluting substances: chemical substances (Atomic economy, Factor E) and energy.

### 2. Use of renewable sources from raw materials

Chemical products from glucose. Chemicals from fatty acids. Polymers from renewable sources. Other products from renewable sources.

### 3. Sustainable Process Design and Industrial Examples

Factors to consider for the design of a sustainable process. Complete study. Industrial examples.

### 4. Renewable energy sources

Main renewable energy sources. Solar, eolic, hydroelectric and biomass. Other renewable energy sources. Energy storage systems.

### 5. Recycling

Waste recycling: paper, plastics, glass, batteries, common metals (Al, Pb,...), scarce metals (Au, Rh, Pd, Ta,...),...

### 6. Environmental pollution: solutions provided by chemistry

Introduction to the main problems of environmental pollution: global warming, photochemical smog, ozone layer and acid rain. Contributions of chemistry to their resolution.



## 7. Catalysis: green concepts and applications

Basic concepts of catalysis. -Catalysis and Green Chemistry. -Influence of catalytic processes in the E-factor and atomic efficiency. -Examples of the role of catalytic processes in Q.V.: Catalysis by acids and bases. Oxidation and catalytic reduction. Catalytic formation and C-C bonds. Enantioselective catalysis.

## 8. Biocatalysis and biotechnology

Biocatalysis. - Renewable materials and white biotechnology. -Integration of processes and catalytic cascades. -Catalysis for a green industry

## WORKLOAD

ACTIVITY	Hours	% To be attended
Theory classes	21,00	100
Tutorials	6,00	100
Seminars	3,00	100
Development of group work	10,00	0
Study and independent work	15,00	0
Readings supplementary material	10,00	0
Preparation of evaluation activities	10,00	0
<b>TOTAL</b>	<b>75,00</b>	

## TEACHING METHODOLOGY

The subject will be taught through participatory lectures, classes with directed practical activity, seminars and workshops where, among other training activities, the resolution of applied practical problems aimed at evaluating the understanding of the subject will be carried out. pupil. In addition, use will be made of the Virtual Classroom platform, a virtual space where all the information deemed appropriate for the development of teaching and the control of student participation in the proposed activities is deposited. Dynamics of continuous evaluation (discussion and network activities, on-line, etc ...).

Due to organizational reasons, during the 2022-2023 academic year, attendance has been reduced to 80%

## EVALUATION

The evaluation of the course will be carried out in a continuous manner by the teacher throughout the course and will consist of the following sections.



- Direct evaluation of the teacher. 20% of the note shall direct evaluation of the professor in theoretical classes and problems and in the tutorials. This assessment shall take into account various aspects, which include:
  - Attendance and participation reasoned and clear in the discussions raised.
  - Problem solving and approach of doubts.
  - Critical thinking.
  - Delivery of exercises.
- Evaluation of seminars. Be taken into account the responses to the questionnaires raised them on seminars. To this section you will be up 30% of the final note.
- Oral exam. 50% of the note is obtained from the oral exam that will be made by the students of the assigned work. The skills such as problem solving, of content related to the matter that will be of such a nature that make the student to relate different aspects that appear on different topics of the course and even in different subjects will both be assessed.

## REFERENCES

### Basic

- M. Lancaster, Green Chemistry, An Introductory Text, Royal Society of Chemistry, Cambridge, 2002
- J. Clark, D. Macquarrie, Handbook of Green Chemistry and Technology, Blackwell, Oxford, 2002
- P. T. Anastas, J. C. Warner, Green Chemistry: Theory and Practice, Oxford University Press, Oxford, 1998
- R. Mestres, Química Sostenible, Ed. Síntesis, 2011
- Rothenberg, G., Catalysis. Concepts and Green Applications. Wiley-VCH, Weinheim, ISBN: 978-3-527-31824-7

### Additional

- M. C. Cann, M. E. Connelly, Real-World Cases in Green Chemistry, American Chemical Society, Washington, 2000





- Revista Green Chemistry, 24 números año, Walter Leitner ed., RSC, desde 1999
- R. L. Garrett, Pollution Prevention, Green Chemistry, and the Design of Safer Chemicals, en, S. C. DeVito y R. L. Garrett Ed., Designing Safer Chemicals, ACS Symposium Series, American Chemical Society, Washington, 1996

